# Quantitative structure-toxicity relationship studies of aromatic aldehydes to Tetrahymena pyriformis based on electronic and topological descriptors 

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#### Abstract

To establish a quantitative structure-toxicity relationship (QSTR) of a series of 77 aromatic aldehydes for their acute toxicity against Tetrahymena pyriformis, were used the principal component analysis, the multiple linear regression and the multiple nonlinear regression analysis. We proposed linear and nonlinear models and interpreted the toxicity of the compounds by multivariate statistical analysis. The proposed models have been validated using internal validation and external validation techniques and an agreement between experimental and predicted values was verified. The applicability domains of MLR models were investigated using William's plot to detect outliers and outsides compounds. It is interesting to note that some of the selected electronic and topological descriptors in our models are better for the prediction of new similar molecules.


## 1. Introduction

With the advent of modern science, technology and industrialization, the use of aromatic aldehydes is getting increased accompanied with an increased number of new chemicals [1]. Many of these chemicals were released into the environment and accumulated in nearly all natural environments, especially in aquatic systems, so it is beneficial to study seriously their potential hazard to aquatic organism.
Experiment is a direct way to obtain the toxicity data of organic compounds, but it has many deficiencies, such as requirement of enormous number of trial organisms, expensive cost, long time, the difference in measured value between different researchers. Consequently, it would be very difficult to obtain the toxicity data of all organic compounds by experiment, as new compounds are springing up, other difficulties will follow. So it is necessary to use the theoretical research to make up for disadvantages of the experiment and to predict the toxicity data of compounds quickly and exactly.
With the rapid development of computational science and theoretical chemistry, it can quickly and precisely obtain the quantum chemical parameters of organic compounds. Quantitative structure-activity relationship (QSAR) can predict the bioactivity such as toxicity, mutagenicity and carcinogenicity based on structural parameters of compounds and appropriate mathematical models.
At present, there are a large number of molecular descriptors that can be used in QSAR studies [2-3]. Once validated, the findings can be used to predict activities of untested compounds.
The aim of this study is to develop predictive QSTR models for the acute toxic effects of aromatic aldehydes towards Tetrahymena pyriformis using several statistical tools, principal components analysis (PCA), multiple linear regression (MLR) and multiple non-linear regression (MNLR).

## 2. Material and methods

### 2.1. Experimental data

To determine a quantitative structure-toxicity relationship, we studied a series of 77 selected aromatic aldehydes for their acute toxicity against the protozoan ciliate Tetrahymena pyriformis [4]. 66 molecules were selected to propose the quantitative model (training set) as well as 11 compounds that were not used in the training set were
selected randomly served to test the performance of the proposed model (test set). The following table shows the studied compounds and the corresponding experimental toxicties $\mathrm{pIC}_{50}$ (Table 1).

Table 1: The range of the toxicity data varies between -1.50 and $2.63(\mu \mathrm{M})$

| $\mathbf{N}^{\circ}$ | Name (IUPAC) | pIC 50 | $\mathbf{N}^{\circ}$ | Name (IUPAC) | pIC 50 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 4-Nitrobenzaldehyde | 0.203 | 40 | 2-Chloro-3-hydroxy-4-methoxybenzaldehyde | 0.204 |
| 2 | 1-Naphthaldehyde | 0.423 | 41 | 6-Chloro-2-fluoro-3-methylbenzaldehyde | 1.238 |
| 3 | 4-Biphenylcarboxaldehyde | 1.119 | 42 | 3-Chloro-2-fluoro-5-(trifluoromethyl)benzaldehyde | 1.723 |
| 4 | 4-Bromobenzaldehyde | 0.587 | 43 | 2,3,5-Trichlorobenzaldehyde | 1.499 |
| 5 | 4-Cyanobenzaldehyde | 0.043 | 44 | 2-Fluorenecarboxaldehyde | 1.499 |
| 6 | Benzaldehyde | -0.196 | 45 | 2-Methyl-1-naphthaldehyde | 1.231 |
| 7 | p-Tolualdehyde | -0.057 | 46 | 4-Methyl-1-naphthaldehyde | 1.123 |
| 8 | 4-Fluorobenzaldehyde | -0.127 | 47 | Phenanthrene-9-carboxaldehyde | 1.708 |
| 9 | 4-Chlorobenzaldehyde | 0.400 | 48 | 5-Hydroxy-2-nitrobenzaldehyde | 0.329 |
| 10 | 4-Ethylbenzaldehyde | 0.291 | 49 | 3-Hydroxy-4-nitrobenzaldehyde | 0.273 |
| 11 | Terephthaldicarboxaldehyde | -0.086 | 50 | 3-Hydroxybenzaldehyde | 0.085 |
| 12 | 4-Anisaldehyde | -0.047 | 51 | 3-Hydroxy-4-methoxybenzaldehyde | -0.142 |
| 13 | 4-Ethoxybenzaldehyde | 0.073 | 52 | 3,4-Dimethoxy-5-hydroxycarboxaldehyde | -0.390 |
| 14 | 4-Acetamidobenzaldehyde | -0.224 | 53 | 2,3-Dihydroxybenzaldehyde | 0.111 |
| 15 | 2-Tolualdehyde | 0.011 | 54 | 2,5-Dihydroxybenzaldehyde | 0.277 |
| 16 | 3-Tolualdehyde | 0.081 | 55 | 3,4-Dihydroxybenzaldehyde | 0.107 |
| 17 | 2-Chlorobenzaldehyde | 0.487 | 56 | 3,4,5-Trihydroxybenzaldehyde | -0.196 |
| 18 | 3-Chlorobenzaldehyde | 0.406 | 57 | 2,3,4-Trihydroxybenzaldehyde | 0.001 |
| 19 | 2-Nitrobenzaldehyde | 0.167 | 58 | 2,4,6-Trihydroxybenzaldehyde | 0.128 |
| 20 | 3-Nitrobenzaldehyde | 0.178 | 59 | 2,4-Dihydroxybenzaldehyde | 0.515 |
| 21 | Phenyl-1,3-dialdehyde | 0.183 | 60 | 3-Ethoxy-2-hydroxycarboxaldehyde | 0.850 |
| 22 | 2-Anisaldehyde | 0.148 | 61 | 3-Methoxysalicylaldehyde | 0.377 |
| 23 | 3-Anisaldehyde | 0.232 | 62 | 3,5-Dibromosalicylaldehyde | 1.648 |
| 24 | 3-Bromobenzaldehyde | 0.506 | 63 | 4,6-Dimethoxy-2-hydroxybenzaldehyde | 0.617 |
| 25 | 3-Fluorobenzaldehyde | 0.154 | 64 | 2-Hydroxy-3-nitrocarboxaldehyde | 0.870 |
| 26 | 2,4-Dichlorobenzaldehyde | 1.036 | 65 | 2-Chloro-4-hydroxycarboxaldehyde | 0.890 |
| 27 | 2,4-Dimethoxybenzaldehyde | -0.056 | 66 | 4-Hydroxy-3-nitrobenzaldehyde | 0.614 |
| 28 | 2,4,5-Trimethoxybenzaldehyde | -0.101 | 67 | 4-Hydroxybenzaldehyde | 0.266 |
| 29 | 4-(Dimethylamino)benzaldehyde | 0.231 | 68 | 2-Hydroxy-1-naphthaldehyde | 1.320 |
| 30 | 4-Phenoxybenzaldehyde | 1.257 | 69 | 5-Bromovanillin | 0.617 |
| 31 | 2-Bromobenzaldehyde | 0.477 | 70 | 4-Hydroxy-1-naphthaldehyde | 1.050 |
| 32 | 2-Fluorobenzaldehyde | 0.079 | 71 | 5-Bromosalicylaldehyde | 1.107 |
| 33 | 4-Butoxybenzaldehyde | 0.716 | 72 | 5-Chlorosalicylaldehyde | 1.009 |
| 34 | 4-(Pentyloxy)benzaldehyde | 1.179 | 73 | 2-Hydroxybenzaldehyde | 0.424 |
| 35 | 4-Isopropylbenzaldehyde | 0.67 | 74 | 3-Bromo-4-hydroxycarboxaldehyde | 0.610 |
| 36 | Pentafluorobenzaldehyde | 0.815 | 75 | 3-Methoxy-4-hydroxybenzaldehyde | -0.030 |
| 37 | 2-Chloro-5-nitrobenzaldehyde | 0.527 | 76 | 3,5-Dibromo-4-hydroxycarboxaldehyde | 0.890 |
| 38 | 2-Chloro-6-fluorobenzaldehyde | 0.155 | 77 | 3-Ethoxy-4-hydroxybenzaldehyde | 0.015 |
| 39 | 3-Cyanobenzaldehyde | -0.020 |  |  |  |

*Test set
2.2. Computational methods

An attempt has been made to correlate the toxicity of these compounds with various physicochemical parameters. DFT (density functional theory) and ChemSketch program methods were used in this study. 3D structures of the molecules were generated using the Gauss View 3.0 and then, all of the calculations were performed using the Gaussian 03 W program series. Geometry optimization of the 77 compounds was carried out by a B3LYP function employing a 6-31G (d) basis set [5,6]. The geometry of all of the species under investigation was determined by optimizing all of the geometrical variables without any symmetry constraints [7].

### 2.3. Calculation of the molecular descriptors

From the results of the DFT calculations, then some related structural descriptors from the results of quantum computation were chosen: the highest occupied molecular orbital energy $\mathrm{E}_{\text {Номо }}(\mathrm{eV})$, the lowest unoccupied molecular orbital energy $\mathrm{E}_{\mathrm{LUMO}}(\mathrm{eV})$, the energy gap $\Delta \mathrm{E}(\mathrm{eV})$, the dipole moment $\mu$ (Debye), the total energy $\mathrm{E}_{\mathrm{T}}$ (eV).

ChemSketch program [8] was employed to calculate the others molecular descriptors such as: the molar volume MV $\left(\mathrm{cm}^{3}\right)$, the molecular weight MW ( $\mathrm{g} / \mathrm{mol}$ ), the molar refractivity MR $\left(\mathrm{cm}^{3}\right)$, the parachor $\mathrm{Pc}\left(\mathrm{cm}^{3}\right)$, the density $\mathrm{D}\left(\mathrm{g} / \mathrm{cm}^{3}\right)$, the refractive Index n , the surface tension $\gamma$ (Dyne/cm) and the polarizability $\alpha\left(\mathrm{cm}^{3}\right)$. To improve the estimate quality of toxicity of these compounds, molecular descriptor which reflect other specific interactions should be also included as octanol/water partition coefficient $(\log P)$.

### 2.4. Statistical analysis

To explain the structure-toxicity relationship, these 14 descriptors were calculated for the 77 molecules using the Gaussian 03W and ChemSketch program software. The study that we conducted consists of multiple linear regression (MLR) and non-linear regression (MNLR), which are available in the XLSTAT software [9]. The multiple linear regression statistical techniques used to study the relationship between one dependent variable and several independent variables. It is a mathematical technique that minimizes the differences between actual and predicted values. It has also served to select descriptors that are used as input parameters in multiple nonlinear regression (MNLR).The MLR and MNLR techniques was employed to model the structure-toxicity relationships. The equations were justified by the correlation coefficient (R), the Mean Squared Error (MSE), the Fisher F-statistic ( F ), and the significance level (p-value) [10].

## 3. Results and discussion

3.1. Data set for analysis

QSTR analysis was performed using the $\mathrm{pIC}_{50}$ of 77 selected aromatic aldehydes to Tetrahymena pyriformis as reported in [4], the values of the 14 chemical descriptors are shown in table 2.

Table 2: The values of the fourteen chemical descriptors

| N | pIC 50 | MW | MR | MV | Pc | n | $\square$ | D | $\square$ | $\mathbf{E}_{\text {T }}$ | $\mathbf{E}_{\text {Hомо }}$ | $\mathbf{E}_{\text {LUMO }}$ | $\square \mathbf{E}$ | $\square$ | $\log P$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $1{ }^{*}$ | 0.203 | 151.12 | 39.55 | 112.90 | 307.80 | 1.62 | 55.10 | 1.34 | 15.67 | -14978.44 | 7.03 | -3.02 | -10.05 | 2.27 | 1.303 |
| 2 | 0.423 | 156.18 | 50.84 | 135.20 | 356.10 | 1.68 | 48.10 | 1.16 | 20.15 | -13593.56 | -6.21 | -1.99 | 4.22 | 3.55 | 2.463 |
| 3 | 1.119 | 182.22 | 57.59 | 166.30 | 425.60 | 1.61 | 42.80 | 1.10 | 22.83 | -15701.67 | -6.43 | -1.85 | 4.57 | 3.86 | 2.777 |
| 4 | 0.587 | 185.02 | 40.69 | 117.20 | 302.80 | 1.61 | 44.40 | 1.58 | 16.13 | -79421.13 | -7.02 | -1.99 | 5.03 | 2.07 | 2.231 |
| 5 | 0.043 | 131.13 | 36.28 | 113.50 | 300.20 | 1.55 | 48.90 | 1.15 | 14.38 | -11921.64 | -7.52 | -2.63 | 4.89 | 2.62 | 1.814 |
| 6 | -0.196 | 106.12 | 33.00 | 101.00 | 252.30 | 1.57 | 38.80 | 1.05 | 13.08 | -9409.96 | -6.95 | -1.71 | 5.24 | 3.30 | 1.343 |
| 7 | -0.057 | 120.15 | 37.83 | 117.30 | 289.90 | 1.56 | 37.20 | 1.02 | 14.99 | -10480.62 | -6.85 | -1.60 | 5.25 | 3.78 | 1.758 |
| 8 | -0.127 | 124.11 | 32.99 | 105.30 | 259.40 | 1.54 | 36.80 | 1.18 | 13.08 | -12112.10 | -7.06 | -1.76 | 5.30 | 2.31 | 1.479 |
| 9 | 0.400 | 140.57 | 37.90 | 113.00 | 288.20 | 1.59 | 42.20 | 1.24 | 15.02 | -21924.75 | -7.16 | -1.99 | 5.17 | 1.98 | 1.962 |
| 10 | 0.291 | 134.18 | 42.55 | 133.90 | 328.90 | 1.55 | 36.30 | 1.00 | 16.87 | -11551.08 | -6.82 | -1.57 | 5.25 | 3.96 | 2.214 |
| 11 | -0.086 | 134.13 | 39.75 | 112.70 | 297.30 | 1.62 | 48.30 | 1.19 | 15.76 | -12495.73 | -7.29 | -2.67 | 4.62 | 1.41 | 0.82 |
| 12 | -0.047 | 136.15 | 39.68 | 125.10 | 309.00 | 1.55 | 37.20 | 1.09 | 15.73 | -12528.49 | -6.36 | -1.41 | 4.95 | 4.02 | 1.474 |
| 13 | 0.073 | 150.17 | 44.31 | 141.60 | 348.70 | 1.54 | 36.80 | 1.06 | 17.56 | -13599.10 | -6.29 | -1.37 | 4.92 | 4.25 | 1.818 |
| 14 | -0.224 | 163.17 | 47.27 | 134.10 | 356.00 | 1.62 | 49.50 | 1.22 | 18.74 | -15088.10 | -6.64 | -3.61 | 3.03 | 3.70 | 0.759 |
| 15 | 0.011 | 120.15 | 37.83 | 117.30 | 289.90 | 1.56 | 37.20 | 1.02 | 14.99 | -10480.51 | -6.83 | -1.71 | 5.12 | 3.34 | 1.758 |
| 16 | 0.081 | 120.15 | 37.83 | 117.30 | 289.90 | 1.56 | 37.20 | 1.02 | 14.99 | -10480.60 | -6.88 | -1.65 | 5.23 | 3.71 | 1.758 |
| 17 | 0.487 | 140.57 | 37.90 | 113.00 | 288.20 | 1.59 | 42.20 | 1.24 | 15.02 | -21924.65 | -7.11 | -2.01 | 5.09 | 3.30 | 1.962 |
| 18 | 0.406 | 140.57 | 37.90 | 113.00 | 288.20 | 1.59 | 42.20 | 1.24 | 15.02 | -21924.73 | -7.13 | -2.03 | 5.10 | 1.79 | 1.962 |
| 19* | 0.167 | 151.12 | 39.55 | 112.90 | 307.80 | 1.62 | 55.10 | 1.34 | 15.67 | -14978.05 | -7.39 | -2.60 | 4.79 | 6.52 | 1.303 |
| 20 | 0.178 | 151.12 | 39.55 | 112.90 | 307.80 | 1.62 | 55.10 | 1.34 | 15.67 | -14978.44 | -7.51 | -2.75 | 4.76 | 5.38 | 1.303 |
| $21 *$ | 0.183 | 134.13 | 39.75 | 112.70 | 297.30 | 1.62 | 48.30 | 1.19 | 15.76 | -12495.71 | -7.14 | -2.16 | 4.99 | 5.41 | 0.82 |
| 22 | 0.148 | 136.15 | 39.68 | 125.10 | 309.00 | 1.55 | 37.20 | 1.09 | 15.73 | -12528.27 | -6.29 | -1.44 | 4.85 | 3.87 | 1.474 |
| 23 | 0.232 | 136.15 | 39.68 | 125.10 | 309.00 | 1.55 | 37.20 | 1.09 | 15.73 | -12528.45 | -6.34 | -1.67 | 4.66 | 2.18 | 1.474 |
| 24 | 0.506 | 185.02 | 40.69 | 117.20 | 302.80 | 1.61 | 44.40 | 1.58 | 16.13 | -79421.11 | -7.00 | -2.03 | 4.97 | 3.57 | 2.231 |
| 25 | 0.154 | 124.11 | 32.99 | 105.30 | 259.40 | 1.54 | 36.80 | 1.18 | 13.08 | -12112.06 | -7.15 | -1.95 | 5.19 | 3.49 | 1.479 |
| 26 | 1.036 | 175.01 | 42.79 | 125.00 | 324.00 | 1.60 | 45.10 | 1.40 | 16.96 | -34439.40 | -7.30 | -2.26 | 5.05 | 1.63 | 2.581 |
| 27 | -0.056 | 166.17 | 46.36 | 149.10 | 365.60 | 1.53 | 36.10 | 1.11 | 18.37 | -15629.12 | -6.13 | -1.08 | 5.05 | 6.36 | 1.605 |
| 28 | -0.101 | 196.20 | 53.04 | 173.10 | 422.30 | 1.52 | 35.40 | 1.13 | 21.02 | -18765.17 | -5.45 | -1.18 | 4.27 | 5.28 | 1.736 |
| 29 | 0.231 | 149.19 | 47.31 | 139.00 | 354.30 | 1.60 | 42.10 | 1.07 | 18.75 | -13058.02 | -5.50 | -1.06 | 4.43 | 6.32 | 2.177 |
| 30 | 1.257 | 198.22 | 59.44 | 171.60 | 443.40 | 1.61 | 44.50 | 1.15 | 23.56 | -17749.58 | -6.36 | -1.49 | 4.87 | 4.70 | 3.318 |
| 31 | 0.477 | 185.02 | 40.69 | 117.20 | 302.80 | 1.61 | 44.40 | 1.58 | 16.13 | -79421.07 | -7.03 | -2.00 | 5.03 | 3.25 | 2.231 |
| 32* | 0.079 | 124.11 | 32.99 | 105.30 | 259.40 | 1.54 | 36.80 | 1.18 | 13.08 | -12112.04 | -7.03 | -1.88 | 5.15 | 3.40 | 1.479 |
| 33 | 0.716 | 178.23 | 53.58 | 174.60 | 428.30 | 1.53 | 36.20 | 1.02 | 21.24 | -15740.19 | -6.33 | -1.36 | 4.97 | 5.13 | 2.73 |
| 34 | 1.179 | 192.25 | 58.21 | 191.10 | 468.10 | 1.52 | 35.90 | 1.01 | 23.07 | -16810.66 | -6.32 | -1.35 | 4.97 | 5.16 | 3.186 |
| 35 | 0.67 | 148.20 | 47.19 | 151.10 | 367.30 | 1.54 | 34.80 | 0.98 | 18.70 | -12621.39 | -2.58 | -1.41 | 1.17 | 4.85 | 2.418 |
| 36 | 0.815 | 196.07 | 32.97 | 122.10 | 287.90 | 1.45 | 30.80 | 1.61 | 13.07 | -22919.39 | -7.44 | -2.41 | 5.04 | 2.00 | 2.572 |
| $37^{*}$ | 0.527 | 185.56 | 44.44 | 124.80 | 343.60 | 1.63 | 57.30 | 1.49 | 17.61 | -27493.09 | -7.64 | -2.95 | 4.69 | 4.33 | 1.922 |
| 38 | 0.155 | 158.56 | 37.89 | 117.20 | 295.30 | 1.56 | 40.20 | 1.35 | 15.02 | -24607.69 | -7.15 | -3.36 | 3.79 | 3.98 | 2.098 |
| $39^{*}$ | -0.020 | 131.13 | 36.28 | 113.50 | 300.20 | 1.55 | 48.90 | 1.15 | 14.38 | -11921.67 | -7.49 | -2.37 | 5.11 | 2.15 | 1.814 |
| 40 | 0.204 | 186.59 | 46.46 | 135.40 | 359.80 | 1.60 | 49.70 | 1.38 | 18.41 | -27091.24 | -6.22 | -2.01 | 4.20 | 5.96 | 1.704 |
| 41 | 1.238 | 172.58 | 42.71 | 133.50 | 332.90 | 1.55 | 38.60 | 1.29 | 16.93 | -25697.21 | -6.85 | -2.01 | 4.84 | 4.82 | 2.513 |
| 42 | 1.723 | 226.56 | 42.87 | 150.70 | 352.50 | 1.48 | 29.80 | 1.50 | 16.99 | -33628.15 | -7.61 | -2.63 | 4.98 | 1.26 | 3.418 |
| 43 | 1.499 | 209.46 | 47.69 | 136.90 | 359.90 | 1.61 | 47.70 | 1.53 | 18.90 | -46953.96 | -7.27 | -2.52 | 4.75 | 1.52 | 3.2 |
| 44 | 1.499 | 194.23 | 60.54 | 159.90 | 431.80 | 1.68 | 53.00 | 1.21 | 24.00 | -16739.62 | -4.13 | -1.91 | 2.22 | 3.71 | 2.801 |
| 45 | 1.231 | 170.21 | 55.67 | 151.40 | 393.80 | 1.66 | 45.60 | 1.12 | 22.07 | -14664.09 | -6.13 | -1.91 | 4.22 | 3.17 | 2.878 |


| 46 | 1.123 | 170.21 | 55.67 | 151.40 | 393.80 | 1.66 | 45.60 | 1.12 | 22.07 | -14664.17 | -6.10 | -1.92 | 4.17 | 3.92 | 2.878 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 47 | 1.708 | 206.24 | 68.69 | 169.30 | 459.90 | 1.75 | 54.40 | 1.22 | 27.23 | -17776.88 | -4.98 | -1.82 | 3.16 | 2.16 | 3.583 |
| 48 | 0.329 | 167.12 | 41.43 | 111.30 | 322.80 | 1.67 | 70.50 | 1.50 | 16.42 | -17026.21 | -7.05 | -2.52 | 4.53 | 7.29 | 0.914 |
| $49^{*}$ | 0.273 | 167.12 | 41.43 | 111.30 | 322.80 | 1.67 | 70.50 | 1.50 | 16.42 | -17026.81 | -7.17 | -3.30 | 3.87 | 1.41 | 0.914 |
| 50 | 0.085 | 122.12 | 34.88 | 99.50 | 267.30 | 1.62 | 52.00 | 1.23 | 13.83 | -11458.09 | -6.43 | -1.69 | 4.75 | 4.17 | 0.954 |
| 51 | -0.142 | 152.15 | 41.56 | 123.50 | 324.00 | 1.59 | 47.30 | 1.23 | 16.47 | -14576.64 | -6.00 | -1.40 | 4.59 | 5.18 | 1.085 |
| 52 | -0.390 | 226.18 | 55.17 | 160.00 | 442.80 | 1.61 | 58.50 | 1.41 | 21.87 | -22828.98 | -4.03 | -2.83 | 1.20 | 8.71 | 0.696 |
| 53 | 0.111 | 138.12 | 36.76 | 97.90 | 282.30 | 1.67 | 69.00 | 1.41 | 14.57 | -13506.24 | -6.13 | -1.51 | 4.62 | 5.72 | 0.565 |
| 54 | 0.277 | 138.12 | 36.76 | 97.90 | 282.30 | 1.67 | 69.00 | 1.41 | 14.57 | -13506.12 | -5.91 | -1.61 | 4.30 | 6.05 | 0.565 |
| 55 | 0.107 | 138.12 | 36.76 | 97.90 | 282.30 | 1.67 | 69.00 | 1.41 | 14.57 | -13506.36 | -6.19 | -1.48 | 4.71 | 2.30 | 0.565 |
| 56 | -0.196 | 154.12 | 38.65 | 96.30 | 297.30 | 1.73 | 90.50 | 1.60 | 15.32 | -15554.32 | -5.97 | -1.49 | 4.48 | 1.15 | 0.176 |
| 57 | 0.001 | 154.12 | 38.65 | 96.30 | 297.30 | 1.73 | 90.50 | 1.60 | 15.32 | -15554.27 | -6.34 | -1.27 | 5.07 | 4.51 | 0.176 |
| 58 | 0.128 | 154.12 | 38.65 | 96.30 | 297.30 | 1.73 | 90.50 | 1.60 | 15.32 | -15554.24 | -6.13 | -1.05 | 5.08 | 5.65 | 0.176 |
| 59* | 0.515 | 138.12 | 36.76 | 97.90 | 282.30 | 1.67 | 69.00 | 1.41 | 14.57 | -13506.30 | -6.31 | -1.32 | 5.00 | 4.57 | 0.565 |
| 60 | 0.850 | 210.18 | 53.12 | 152.50 | 425.90 | 1.61 | 60.70 | 1.38 | 21.06 | -20781.76 | -6.86 | -2.39 | 4.47 | 5.50 | 1.160 |
| 61 | 0.377 | 152.15 | 41.56 | 123.50 | 324.00 | 1.59 | 47.30 | 1.23 | 16.47 | -14576.58 | -6.02 | -1.43 | 4.59 | 6.16 | 1.085 |
| 62 | 1.648 | 279.91 | 50.26 | 131.80 | 368.30 | 1.69 | 60.80 | 2.12 | 19.92 | -151480.4 | -6.66 | -2.15 | 4.51 | 3.56 | 2.73 |
| $63^{*}$ | 0.617 | 182.17 | 48.24 | 147.50 | 380.60 | 1.57 | 44.30 | 1.23 | 19.12 | -17695.53 | -6.07 | -1.37 | 4.70 | 4.66 | 1.216 |
| 64 | 0.870 | 211.13 | 48.36 | 123.90 | 384.90 | 1.71 | 93.10 | 1.70 | 19.17 | -17027.11 | -6.45 | -3.13 | 3.32 | 3.94 | 0.271 |
| 65 | 0.890 | 200.58 | 46.71 | 124.00 | 365.30 | 1.68 | 75.30 | 1.62 | 18.51 | -29106.95 | -7.05 | -2.03 | 5.02 | 5.34 | 1.507 |
| 66 | 0.614 | 167.12 | 41.43 | 111.30 | 322.80 | 1.67 | 70.50 | 1.50 | 16.42 | -17026.92 | -7.26 | -3.09 | 4.17 | 1.09 | 0.914 |
| 67 | 0.266 | 122.12 | 34.88 | 99.50 | 267.30 | 1.62 | 52.00 | 1.23 | 13.83 | -11458.17 | -6.50 | -1.45 | 5.04 | 3.38 | 0.954 |
| 68 | 1.320 | 172.18 | 52.72 | 133.60 | 371.10 | 1.72 | 59.40 | 1.29 | 20.90 | -15641.51 | -5.96 | -1.78 | 4.19 | 4.43 | 2.074 |
| 69 | 0.617 | 231.04 | 49.25 | 139.70 | 374.50 | 1.62 | 51.60 | 1.65 | 19.52 | -84587.80 | -6.31 | -1.65 | 4.66 | 3.07 | 1.973 |
| 70 | 1.050 | 172.18 | 52.72 | 133.60 | 371.10 | 1.72 | 59.40 | 1.29 | 20.90 | -15641.78 | -5.88 | -1.74 | 4.14 | 3.66 | 2.074 |
| 71 | 1.107 | 201.02 | 42.57 | 115.70 | 317.80 | 1.66 | 56.90 | 1.74 | 16.87 | -81469.21 | -6.47 | -1.90 | 4.57 | 5.11 | 1.842 |
| 72 | 1.009 | 156.57 | 39.78 | 111.40 | 303.20 | 1.63 | 54.70 | 1.40 | 15.77 | -23972.82 | -6.55 | -1.91 | 4.64 | 5.19 | 1.573 |
| 73 | 0.424 | 122.12 | 34.88 | 99.50 | 267.30 | 1.62 | 52.00 | 1.23 | 13.83 | -11458.08 | -6.50 | -1.58 | 4.92 | 4.77 | 0.954 |
| 74 | 0.610 | 201.02 | 42.57 | 115.70 | 317.80 | 1.66 | 56.90 | 1.74 | 16.87 | -81469.42 | -6.69 | -1.73 | 4.96 | 2.90 | 1.842 |
| 75 | -0.030 | 152.15 | 41.56 | 123.50 | 324.00 | 1.59 | 47.30 | 1.23 | 16.47 | -14576.66 | -6.05 | -1.39 | 4.65 | 5.05 | 1.085 |
| 76 | 0.890 | 279.91 | 50.26 | 131.80 | 368.30 | 1.69 | 60.80 | 2.12 | 19.92 | -151480.4 | -6.83 | -1.96 | 4.87 | 3.50 | 2.73 |
| 77 | 0.015 | 166.17 | 46.19 | 140.00 | 363.70 | 1.57 | 45.50 | 1.19 | 18.31 | -15647.07 | -6.34 | -1.44 | 4.90 | 5.13 | 1.429 |

${ }^{*}$ Test set
3.2. Principal component analysis

The principal component analysis (PCA) was performed to the 14 descriptors of 77 molecules. The first three axes F1, F2 and F3 represent respectively ( $35.50 \% ; 20.64 \%$ and $14.92 \%$ ) of the total variance and they estimate $71.06 \%$ of the total information.
The principal component analysis (PCA) [11] was conducted to identify the link between the different descriptors. Bold values are different from 0 at a significance level of $\mathrm{p}=0.05$. The Pearson correlation coefficients are listed in table 3. The obtained matrix provides information on the positive or negative correlation between descriptors. In general, the co-linearity ( $r>0.5$ ) was observed between most of the variables, and between the variables and $\mathrm{pIC}_{50}$. Additionally, to decrease the redundancy presented in our data matrix, the descriptors that are highly correlated $(\mathrm{R} \geq 0.9)$, were removed.

Table 3: Correlation matrix between different obtained descriptors

|  | $\mathrm{pIC}_{50}$ | MW | MR | MV | Pc | N | $\square$ | D | $\square$ | $\mathbf{E}_{T}$ | $\mathbf{E}_{\mathrm{HOMO}}$ | $\mathbf{E}_{\text {LUMO }}$ | $\square \mathbf{E}$ | $\square$ | $\log P$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{pIC}_{50}$ | 1 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| MW | 0.622 | 1 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| MR | 0.591 | 0.639 | 1 |  |  |  |  |  |  |  |  |  |  |  |  |
| MV | 0.474 | 0.534 | 0.874 | 1 |  |  |  |  |  |  |  |  |  |  |  |
| Pc | 0.527 | 0.665 | 0.965 | 0.929 | 1 |  |  |  |  |  |  |  |  |  |  |
| N | 0.229 | 0.239 | 0.264 | -0.228 | 0.109 | 1 |  |  |  |  |  |  |  |  |  |
| $\gamma$ | 0.001 | 0.182 | 0.006 | -0.383 | -0.027 | 0.847 | 1 |  |  |  |  |  |  |  |  |
| D | 0.298 | 0.668 | -0.025 | -0.262 | -0.040 | 0.507 | 0.591 | 1 |  |  |  |  |  |  |  |
| $\alpha$ | 0.591 | 0.639 | 1.000 | 0.874 | 0.965 | 0.264 | 0.006 | -0.025 | 1 |  |  |  |  |  |  |
| $\mathbf{E}_{T}$ | -0.367 | -0.711 | -0.143 | -0.038 | -0.099 | -0.203 | -0.083 | -0.763 | -0.143 | 1 |  |  |  |  |  |
| $\mathbf{E}_{\text {Homo }}$ | -0.026 | 0.012 | 0.175 | 0.131 | 0.168 | 0.095 | 0.059 | -0.090 | 0.175 | 0.102 | 1 |  |  |  |  |
| $\mathbf{E}_{\text {LUMO }}$ | -0.078 | -0.214 | 0.037 | 0.100 | 0.012 | -0.073 | -0.148 | -0.295 | 0.037 | 0.084 | 0.003 | 1 |  |  |  |
| $\Delta \mathbf{E}$ | 0.000 | -0.078 | -0.154 | -0.093 | -0.156 | -0.113 | -0.103 | -0.008 | -0.154 | -0.071 | -0.949 | 0.313 | 1 |  |  |
| $\mu$ | -0.179 | 0.048 | 0.192 | 0.198 | 0.258 | 0.029 | 0.105 | -0.103 | 0.191 | 0.134 | 0.067 | 0.265 | 0.020 | 1 |  |
| $\log P$ | 0.699 | 0.436 | 0.542 | 0.656 | 0.497 | -0.271 | -0.570 | -0.094 | 0.542 | -0.318 | -0.042 | 0.018 | 0.045 | -0.243 | 1 |

[^0]Based on the 11 remaining descriptors, a mathematical linear model was proposed to predict quantitatively the physicochemical effects of substituents on the toxicity of the 66 molecules by using backward selection and stepwise selection in the multiple regression analysis.
The study of the descendant MLR multiple linear regression based on the elimination of descriptors until a valid model was obtained and the stepwise multiple linear regression procedures based on the forward selection and backward elimination methods were employed to determine the best regression models.
The QSAR models built using descendant and stepwise multiple linear regression methods are represented by the following equations:
For the descendant MLR:

$$
\begin{equation*}
\mathbf{p I C} \mathbf{5 0}=-8.176+9.64910^{-03} \mathbf{M W}-2.64810^{-02} \mathbf{M R}+4.591 \mathbf{n}+7.22110^{-06} \mathbf{E}_{\mathbf{T}}+0.598 \log \boldsymbol{P} \tag{Equation1}
\end{equation*}
$$



Figure 1: Graphical representation of calculated and observed toxicity by descendant MLR
For our 66 compounds, the correlation between experimental and calculated toxicity based on this model is quite significant (Figure 1) as indicated by statistical values:

$$
N=66 \quad R^{2}=0.799 \quad R_{C V}^{2}=0.743 \quad \text { MSE }=0.064 \quad F=47.672 \quad \text { p-value }<0.0001
$$

The elaborated QSTR model reveals that the toxicity of 66 aromatic aldehydes to Tetrahymena pyriformis could be explained by a number of electronic factors $\left(\mathbf{M W}, \mathbf{n}, \mathbf{E}_{\mathbf{T}}\right.$ and $\left.\log \boldsymbol{P}\right)$. The positive correlation of these factors with the value of the $\mathbf{p I C}_{\mathbf{5 0}}$ in equation 1 shows that an increase in the values of these factors implies an increase in the value of the $\mathbf{p I C}_{\mathbf{5 0}}$, whereas a negative correlation of the $\mathbf{M R}$ shows that an increase in the value of this factor indicates a decrease in the value of the $\mathbf{p I C}_{\mathbf{5 0}}$. For the stepwise MLR:

$$
\mathbf{p I C} \mathbf{5 0}_{\mathbf{0}}=-1.928+0.024 \gamma+0.709 \log \boldsymbol{P} \quad(\text { Equation } 2)
$$

For our 66 compounds, the correlation between experimental toxicity and calculated on based on this model is quite significant (Figure 2) as indicated by statistical values:

$$
\mathrm{N}=66 \quad \mathrm{R}^{2}=0.760 \quad \mathrm{R}_{\mathrm{CV}}^{2}=0.732 \quad \mathrm{MSE}=0.073 \quad \mathrm{~F}=99.483 \quad \text { p-value }<0.0001
$$

The elaborated QSTR model reveals that the toxicity of 66 aromatic aldehydes to Tetrahymena pyriformis may be explained by the two selected descriptors in equation 2 . The positive correlation of the $\gamma$ and $\boldsymbol{\operatorname { l o g } \boldsymbol { P } \text { with the }}$ $\mathbf{p I C}_{50}$ shows that an increase in the values of these factors implies a increase in the value of the $\mathbf{p I C}_{\mathbf{5 0}}$.
The figures 1 and 2 show a very regular distribution of toxicity values depending on the experimental values. In the equation, N is the number of compounds, $\mathrm{R}^{2}$ is the determination coefficient, MSE is the mean squared error, $F$ is the Fisher's criterion and $p$-value is the significance level.
A higher correlation coefficient and lower mean squared error indicate that the model is more reliable. A P that is smaller than 0.05 exhibits that the regression equation is statistically significant. The QSTR models expressed by equation 1 and equation 2 are cross-validated by its noticeable $R^{2}{ }_{c v}$ values $\left(R^{2}{ }_{c v}=0.743\right.$ to a descendant MLR model and $\mathrm{R}_{\mathrm{cv}}=0,732$ to a stepwise MLR model) obtained by the leave-one-out (LOO) method. A value of $R_{c v}^{2}$ is greater than 0.5 is the important criterion for qualifying a QSTR model as valid [12]. The correlation coefficients between descriptors in the descendant MLR model were calculated by variance inflation factor (VIF) as shown in table 4.


Figure 2: Graphical representation of calculated and observed toxicity by stepwise MLR
Table 4: The variance inflation factors (VIF) of descriptors in QSAR model

| Statistique | MW | MR | $\mathbf{n}$ | $\mathbf{E}_{\mathbf{T}}$ | $\boldsymbol{\operatorname { l o g } \boldsymbol { P }}$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| Tolérance | 0.216 | 0.243 | 0.573 | 0.311 | 0.427 |
| VIF | 4.628 | 4.110 | 1.746 | 3.221 | 2.344 |

The VIF was defined as $1 /(1-\mathrm{R} 2)$, where R was the multiple correlation coefficients for one independent variable against all the other descriptors in the model. If VIF greater than 5 , it mean that models were unstable and must be rejected, models with a VIF values between 1 and 5 can be accepted. As can be seen from table 4, the VIF values of the two descriptors are all smaller than 5.0, resulting that there is no-collinearity between the selected descriptors and the obtained model has good stability. With the MLR models, the values of predicted $\mathrm{pIC}_{50}$ calculated from equation 1 and equation 2 and the observed values are given in table 6.

### 3.4. Multiple nonlinear regression (MNLR)

We have used also the technique of nonlinear regression model to improve the structure-toxicity relationship to quantitatively evaluate the effect of the substituents and they have applied to the data matrix constituted obviously from the descriptors proposed by MLR corresponding to the 66 molecules (Training set).
The coefficients $\mathrm{R}^{2}$, MSE are used to select the best regression performance. We used a pre-programmed function of XLSTAT following:

$$
Y=a+(b X 1+c X 2+d X 3+e X 4 \ldots)
$$

Where $\mathrm{a}, \mathrm{b}, \mathrm{c}, \mathrm{d} . .$. : represent the parameters and $\mathrm{X} 1, \mathrm{X} 2, \mathrm{X} 3, \mathrm{X} 4 \ldots$. represent the variables.
The proposed descriptors in equation 1 and equation 2 by MLR models are used as the input parameters in the MNLR method. The QSTR models built using multiple non-linear regression method are represented by the following equations:
The MNLR model using selected descriptors by descendant selection:

$$
\begin{gathered}
\mathbf{p I C}_{\mathbf{5 0}}=-22.973+7.39410^{-03} \mathbf{M W}+2.92310^{-02} \mathbf{M R}+21.783 \mathbf{n}+1.29510^{-05} \mathbf{E}_{\mathbf{T}}+0.456 \log \boldsymbol{P}+7.25010^{-06} \\
\mathbf{M W}^{2}-6.32810^{-04} \mathbf{M R}^{2}-5.298 \mathbf{n}^{2}+3.48910^{-11} \mathbf{E}_{\mathbf{T}}^{2}+0.053(\log \boldsymbol{P})^{2}(\text { (Equation 3) } \\
\mathbf{N}=\mathbf{6 6} \quad \mathbf{R}^{2}=\mathbf{0 . 8 1 0} \quad \mathbf{R}_{\mathbf{c v}}^{2}=\mathbf{0 . 7 1 3} \quad \mathbf{M S E}=\mathbf{0 . 0 6 6}
\end{gathered}
$$

The MNLR model using selected descriptors by stepwise selection:

$$
\begin{gathered}
\mathbf{p I C}_{50}=-2.093+4.111 \quad 10^{-02} \gamma+0.423 \log \boldsymbol{P}-1.70610^{-04} \boldsymbol{\gamma}^{2}+7.15010^{-02}(\log \boldsymbol{P})^{\mathbf{2}} \quad \text { (Equation 4) } \\
\mathbf{N}=\mathbf{6 6} \quad \mathbf{R}^{2}=\mathbf{0 . 7 6 8} \quad \mathbf{R}_{\mathbf{C V}}^{\mathbf{2}}=\mathbf{0 . 7 3 2} \quad \mathbf{M S E}=\mathbf{0 . 0 7 3}
\end{gathered}
$$

The higher values of $R^{2}$ of two MNLR models and the lower mean squared errors MSE indicate that the two proposed models are predictive and reliable. The obtained models were internally validated by the leave-one-out cross-validation technique. The values of $\mathrm{R}^{2}{ }_{\mathrm{cv}}$ for two MNLR models are higher than 0.5 , indicate the better predictivity of MNLR models. The toxicity values $\mathrm{pIC}_{50}$ predicted by this model are almost similar to that observed. The correlations of predicted and observed $\mathrm{pIC}_{50}$ values are illustrated in figure 3.


Figure 3: Graphical representation of calculated and observed toxicity by MNLR ((a): proposed descriptors by descendant selection and (b): by stepwise selection)

### 3.5. External validation

To estimate the predictive power of developed models, we must use a set of compounds that have not been used for training set to establish the QSTR models. The established models in the computation procedure using the 66 aromatic aldehydes are used to predict the toxicity of the remaining 11 compounds. The comparison of the values of $\mathrm{pIC}_{50}$-test and $\mathrm{pIC}_{50}$-obs shows that a good prediction has been obtained for the 11 compounds $\left(\mathrm{R}_{\text {test }}\right.$ and $R_{\text {test }}^{2}$ showed in table 5).

Table 5: Performance comparison between obtained models by the MLR and RNLM

| Model | Training set |  |  |  | Test set |  |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{R}^{\mathbf{2}}$ | $\mathbf{R}^{\mathbf{2} \mathbf{c v}}$ | MSE | R ext | $\mathbf{R}^{\mathbf{2}} \mathbf{e x t}$ | MSE |
| MLR $_{\text {descendant }}$ | 0.799 | 0,743 | 0.064 | 0.852 | 0.726 | 0.212 |
| MLR $_{\text {stepwise }}$ | 0.760 | 0,732 | 0.073 | 0.846 | 0.716 | 0.144 |
| MNLR $_{\text {descendant }}$ | 0.810 | 0.713 | 0.066 | 0.840 | 0.707 | 0.299 |
| MNLR $_{\text {stepwise }}$ | 0.768 | 0.732 | 0.073 | 0.875 | 0.766 | 0,139 |

The true predictive power of these models can be tested from their ability to predict perfectly the $\mathrm{pIC}_{50}$ of compounds from an external test set. The activities of the remaining set of 11 compounds are deduced from the quantitative proposed models in training set. The observed and calculated $\mathrm{pIC}_{50}$ values are given in table 6 . These models were able to predict the activities of test set molecules in agreement with the experimentally determined value. The higher values of $\mathrm{R}^{2}$ test $\left(\mathrm{R}^{2}\right.$ test $=0.726$ for the descendant MLR model, $\mathrm{R}^{2}$ test $=0.716$ for the stepwise MLR model, $\mathrm{R}^{2}$ test $=0.707$ for MNLR model (with descriptors proposed by descendant MLR), and $\mathrm{R}^{2}$ test $=0.766$ for MNLR model (with descriptors proposed by descendant MLR)) indicate the improved predictivity of these models.
A comparison of the quality of MLR and MNLR models shows that four approaches have the good predictive capability; which is sufficient to conclude the performance of these models and to establish a satisfactory relationship between selected descriptors and toxicity. Furthermore, the results obtained by MNLR are relatively better than those obtained by MLR, but the latter approach is more transparent and gives the most interpretable results and a good explanation of the descriptors associated with toxicities.

### 3.6. Domain of applicability

To estimate the reliability of any QSTR model and its ability to predict new compounds, the domain of applicability must be essentially defined [13]. The predicted compounds that fall within this domain may be considered as reliable. The applicability domain was discussed with the Williams graph in figures 4 and 5, which the standardized residuals and the leverage values $\left(h_{i}\right)$ are plotted. It is based on the calculation of the leverage $h_{i}$ for each molecule, for which QSAR model is used to predict its toxicity:

$$
\begin{equation*}
h_{i}=x_{i}\left(X^{T} X\right)^{-1} x_{i}^{T} \quad i=1, \ldots \mathrm{n} \tag{3}
\end{equation*}
$$

where $x_{i}$ is the row vector of the descriptors of compound i and X is the variable matrix deduced from the training set variable values. The index T refers to the matrix/vector transposed. The critical leverage $h^{*}$ is
generally fixed at $3(\mathrm{k}+1) / \mathrm{N}$, where N is the number of training molecules, and k is the number of model descriptors.

Table 6: Observed and predicted values of $\mathrm{pIC}_{50}$ according to different methods

| pIC50 (calc.) |  |  |  |  |  |  |  | pIC50 (calc.) |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{N}^{\circ}$ | $\begin{aligned} & \mathbf{p I C}_{50} \\ & \text { (obs.) } \end{aligned}$ | $\mathbf{M L R}_{\text {step }}$ | $\mathbf{N M L R}_{\text {step }}$ | $\mathbf{M L R}_{\text {desc }}$ | $\mathbf{N M L R}_{\text {desc }}$ | $\mathbf{N}^{\circ}$ | $\begin{aligned} & \mathrm{pIC}_{50} \\ & \text { (obs.) } \end{aligned}$ | $\mathbf{M L R}_{\text {step }}$ | $\mathbf{N M L R}_{\text {step }}$ | $\mathbf{M L R}_{\text {desc }}$ | $\mathbf{N M L R}_{\text {desc }}$ |
| 1 | 0.203 | 0.316 | 0.327 | 0.331 | 0.346 | 40 | 0.204 | 0.472 | 0.457 | 0.569 | 0.554 |
| 2* | 0.423 | 0.972 | 0.966 | 1.051 | 1.111 | 41 | 1.238 | 0.780 | 0.754 | 0.802 | 0.834 |
| 3 | 1.119 | 1.068 | 1.080 | 0.988 | 1.013 | 42 | 1.723 | 1.211 | 1.262 | 1.473 | 1.588 |
| 4 | 0.587 | 0.719 | 0.696 | 0.686 | 0.598 | 43 | 1.499 | 1.485 | 1.566 | 1.564 | 1.677 |
| 5 | 0.043 | 0.531 | 0.512 | 0.254 | 0.250 | 44 | 1.499 | 1.329 | 1.353 | 1.368 | 1.325 |
| 6 | -0.196 | -0.046 | -0.058 | -0.100 | -0.121 | 45 | 1.231 | 1.207 | 1.237 | 1.212 | 1.280 |
| 7 | -0.057 | 0.210 | 0.165 | 0.107 | 0.129 | 46 | 1.123 | 1.207 | 1.237 | 1.212 | 1.280 |
| 8 | -0.127 | 0.003 | -0.029 | 0.012 | -0.052 | 47 | 1.708 | 1.918 | 2.073 | 2.023 | 1.864 |
| 9 | 0.400 | 0.475 | 0.443 | 0.470 | 0.459 | 48* | 0.329 | 0.409 | 0.404 | 0.413 | 0.427 |
| 10 | 0.291 | 0.513 | 0.462 | 0.341 | 0.400 | 49 | 0.273 | 0.409 | 0.404 | 0.413 | 0.427 |
| 11 | -0.086 | -0.189 | -0.110 | -0.086 | -0.041 | 50 | 0.085 | -0.006 | 0.052 | -0.004 | 0.003 |
| 12 | -0.047 | 0.009 | -0.021 | -0.018 | -0.017 | 51 | -0.142 | -0.025 | 0.013 | 0.022 | 0.045 |
| 13 | 0.073 | 0.243 | 0.194 | 0.151 | 0.161 | 52 | -0.390 | -0.033 | 0.057 | 0.167 | 0.137 |
| 14 | -0.224 | -0.204 | -0.114 | -0.060 | -0.022 | 53 | 0.111 | 0.126 | 0.193 | 0.110 | 0.131 |
| 15 | 0.011 | 0.210 | 0.165 | 0.107 | 0.129 | 54 | 0.277 | 0.126 | 0.193 | 0.110 | 0.131 |
| 16* | 0.081 | 0.210 | 0.165 | 0.107 | 0.129 | 55 | 0.107 | 0.126 | 0.193 | 0.110 | 0.131 |
| 17* | 0.487 | 0.475 | 0.443 | 0.470 | 0.459 | 56 | -0.196 | 0.365 | 0.307 | 0.243 | 0.254 |
| 18 | 0.406 | 0.475 | 0.443 | 0.470 | 0.459 | 57 | 0.001 | 0.365 | 0.307 | 0.243 | 0.254 |
| 19 | 0.167 | 0.316 | 0.327 | 0.331 | 0.346 | 58* | 0.128 | 0.365 | 0.307 | 0.243 | 0.254 |
| 20 | 0.178 | 0.316 | 0.327 | 0.331 | 0.346 | 59 | 0.515 | 0.126 | 0.193 | 0.110 | 0.131 |
| 21 | 0.183 | -0.189 | -0.110 | -0.086 | -0.041 | 60 | 0.850 | 0.349 | 0.361 | 0.396 | 0.368 |
| 22 | 0.148 | 0.009 | -0.021 | -0.018 | -0.017 | 61* | 0.377 | -0.025 | 0.013 | 0.022 | 0.045 |
| 23 | 0.232 | 0.009 | -0.021 | -0.018 | -0.017 | 62 | 1.648 | 1.466 | 1.464 | 1.479 | 1.688 |
| 24* | 0.506 | 0.719 | 0.696 | 0.686 | 0.598 | 63 | 0.617 | -0.004 | 0.014 | 0.099 | 0.093 |
| 25* | 0.154 | 0.003 | -0.029 | 0.012 | -0.052 | 64 | 0.870 | 0.494 | 0.376 | 0.467 | 0.516 |
| 26 | 1.036 | 0.984 | 0.982 | 1.022 | 1.055 | 65 | 0.890 | 0.945 | 0.835 | 0.914 | 0.879 |
| 27 | -0.056 | 0.076 | 0.032 | 0.091 | 0.076 | 66 | 0.614 | 0.409 | 0.404 | 0.413 | 0.427 |
| 28 | -0.101 | 0.152 | 0.098 | 0.213 | 0.142 | 67 | 0.266 | -0.006 | 0.052 | -0.004 | 0.003 |
| 29* | 0.231 | 0.625 | 0.595 | 0.543 | 0.608 | 68 | 1.320 | 0.967 | 0.932 | 1.110 | 1.070 |
| 30* | 1.257 | 1.492 | 1.590 | 1.407 | 1.497 | 69 | 0.617 | 0.708 | 0.687 | 0.766 | 0.684 |
| 31 | 0.477 | 0.719 | 0.696 | 0.686 | 0.598 | 70 | 1.050 | 0.967 | 0.932 | 1.110 | 1.070 |
| 32 | 0.079 | 0.003 | -0.029 | 0.012 | -0.052 | 71 | 1.107 | 0.742 | 0.716 | 0.758 | 0.650 |
| 33 | 0.716 | 0.876 | 0.860 | 0.647 | 0.672 | 72 | 1.009 | 0.498 | 0.488 | 0.543 | 0.523 |
| 34 | 1.179 | 1.193 | 1.237 | 0.906 | 0.938 | 73 | 0.424 | -0.006 | 0.052 | -0.004 | 0.003 |
| 35 | 0.670 | 0.621 | 0.572 | 0.413 | 0.470 | 74 | 0.610 | 0.742 | 0.716 | 0.758 | 0.650 |
| 36 | 0.815 | 0.634 | 0.572 | 0.883 | 0.739 | 75 | -0.030 | -0.025 | 0.013 | 0.022 | 0.045 |
| 37 | 0.527 | 0.808 | 0.780 | 0.873 | 0.873 | 76* | 0.890 | 1.466 | 1.464 | 1.479 | 1.688 |
| 38 | 0.155 | 0.524 | 0.486 | 0.586 | 0.559 | 77 | 0.015 | 0.176 | 0.175 | 0.169 | 0.179 |
| 39 | -0.020 | 0.531 | 0.512 | 0.254 | 0.250 |  |  |  |  |  |  |

*Test set
If the leverage value $h$ of molecule is higher than the critical value $\left(h^{*}\right)$ i.e., $h>h^{*}$, the prediction of the compound can be considered as not reliable. From figure 4, five compounds are identified as outliers and one compound among five outliers is considered as outside for the descendant MLR model, which represents $6.49 \%$ of the total of studied compounds. Therefore, the predicted toxicity by the developed MLR model is reliable. The Williams plot for the stepwise MLR model is shown in figure 5.
From this plot, the leverage values $\left(h_{i}\right)$ of any compound in the training and test sets are less than the critical value ( $h^{*}=0.136$ ) excepting the compounds 40 and 54 . Also, the standardized residuals of all compounds in the training and test sets are less than three standard deviation units $( \pm 3 \sigma)$. Therefore, the predicted toxicity by the developed stepwise MLR model is reliable.

### 3.7. Proposed novel compounds

Consequently, with MLR ${ }_{\text {descendant }}$ and MLR $_{\text {stepwise }}$ approach, we can design new compounds with different and improved values of toxicity than the studied compounds. Taking into account the above results, we added suitable substitutions and then calculated the toxicities of the new compounds using the proposed model in equations 1 and 2. The leviers $h$ of new compounds $X_{1}, X_{6}, X_{9}$ and $X_{10}$ for the stepwise model and descendant model are defined as outliers, because they have a higher leverage which is greater than $h^{*}$ ( 0.272 for descendant model and 0.136 for the stepwise model). We can suggest for the six remaining are regarded reliable compounds for design new compounds with different and improved values of toxicity than the studied compounds.


Figure 4: Williams plot for the descendant MLR model (with $h^{*}=0.272$ and residual limits $= \pm 3 \sigma$ )


Figure 5: Williams plot for the stepwise MLR model (with $h^{*}=0.136$ and residual limits $= \pm 3 \sigma$ )

Table 7: Proposed compounds, value of calculated descriptors, and predicted values of $\mathrm{pIC}_{50}$ using $\mathrm{MLR}_{\text {stepwise }}$ model


|  | $\mathbf{2}$ | $\mathbf{3}$ | $\mathbf{4}$ | $\mathbf{5}$ | $\mathbf{6}$ | $\mathbf{E}_{\mathbf{T}}$ | $\boldsymbol{\gamma}$ | $\mathbf{M W}$ | $\mathbf{M R}$ | $\mathbf{n}$ | $\mathbf{L o g} \boldsymbol{P}$ | $\mathbf{p I C}$ <br> $\mathbf{5 0}$ <br> $\mathbf{R L M}$ <br> (step) | $\mathbf{h}$ | $\mathbf{p I C}$ <br> $\mathbf{R L M}$ <br> (desc) |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{X}_{\mathbf{1}}$ | $\mathrm{NH}_{2}$ | $\mathrm{NH}_{2}$ | H | H | $\mathrm{NH}_{2}$ | -511.63 | 87.2 | 151.16 | 45.71 | 1.789 | -1.096 | -0.617 | 0.199 | -0.373 |
| $\mathbf{X}_{\mathbf{2}}$ | $\mathrm{NH}_{2}$ | H | $\mathrm{NH}_{2}$ | H | H | -456.28 | 68.5 | 136.15 | 41.47 | 1.714 | -0.286 | -0.490 | 0.087 | -0.265 |
| $\mathbf{X}_{\mathbf{3}}$ | $\mathrm{NH}_{2}$ | CH 3 | $\mathrm{NH}_{2}$ | CH | CH 3 | -574.23 | 53 | 178.23 | 55.95 | 1.644 | 0.962 | 0.024 | 0.035 | 0.180 |
| $\mathbf{X}_{\mathbf{4}}$ | F | F | H | F | F | -742.47 | 32.1 | 178.08 | 32.98 | 1.471 | 2.413 | 0.553 | 0.033 | 0.859 |
| $\mathbf{X}_{\mathbf{5}}$ | $\mathrm{NO}_{2}$ | H | $\mathrm{NO}_{2}$ | H | H | -754.55 | 71.8 | 196.11 | 46.09 | 1.66 | 1.263 | 0.688 | 0.036 | 0.866 |
| $\mathbf{X}_{\mathbf{6}}$ | OH | OH | OH | OH | H | -646.44 | 117.8 | 170.11 | 40.53 | 1.799 | -0.213 | 0.743 | 0.233 | 0.519 |
| $\mathbf{X}_{\mathbf{7}}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | H | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | -502.83 | 34.5 | 162.22 | 52.3 | 1.541 | 3.003 | 1.029 | 0.056 | 0.871 |
| $\mathbf{X}_{\mathbf{8}}$ | Cl | Cl | Cl | Cl | H | -2183.92 | 49.9 | 243.90 | 52.58 | 1.624 | 3.819 | 1.977 | 0.083 | 2.508 |
| $\mathbf{X}_{\mathbf{9}}$ | H | $\mathrm{C}_{\mathbf{6}} \mathrm{H}_{5}$ | H | $\mathrm{C}_{6} \mathrm{H}_{5}$ | H | -807.68 | 44.7 | 258.31 | 82.18 | 1.627 | 4.211 | 2.131 | 0.116 | 2.122 |
| $\mathbf{X}_{\mathbf{1 0}}$ | Br | Br | Br | Br | H | -10629.9 | 56.3 | 421.70 | 63.76 | 1.695 | 5.097 | 3.037 | 0.166 | 4.957 |

## Conclusion

In this study two different modeling methods, multiple linear regression (MLR) and multiple non linear regression (MNLR), were used for predicting the toxicity of aromatic aldehydes to Tetrahymena pyriformis. The accuracy and predictability of the proposed models were proven by the comparison of key statistical terms of models. The good results obtained with the internal and external validations show that the proposed models in this paper are able to predict activities with a great performance and that the selected descriptors are pertinent. The applicability domains (AD) of the MLR models were defined.
The resulting models have shown that we have established a relationship between some descriptors and the activities in satisfactory manners. The MNLR results have substantially better predictive capability than the MLR results, but the latter gives the most important interpretable results.
The selected descriptors in the QSAR models can illustrate the contributing electronic and steric properties that are responsible for the toxicity of aromatic aldehydes to Tetrahymena pyriformis. By interpreting the molecular descriptors for the stepwise MLR model, we conclude that the increase octanol/water partition coefficient (log P ) and $\gamma$ as well are responsible for the greater activity of the studied compounds, presence of electronegative substituents (like $\mathrm{O}, \mathrm{N}, \mathrm{F}, \mathrm{Br}, \mathrm{Cl}$ ), lipophilic substituents, e.g., chlorine. The aldehydic oxygen was also important for toxicity.
Finally, the accuracy and predictability of the proposed models were illustrated by comparing key statistical indicators such as shown in table 6, the models reported here may be used more conveniently than the previously reported models, with better confidence of prediction accuracy.

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[^0]:    3.3. Multiple linear regressions MLR

